

Effect of Zinc Oxide Percent on Transmission Properties for (X-Ray) of Rubber Compound (SBR)

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Abstract

X-ray beams are comprised of photons of many different energies, and as such are polyenergetic. A characteristic of an x-ray beam that affects the rate of attenuation, and hence the exit intensity of the beam, is its effective energy. Effective energy of the x-ray beam can be considered as the weighted mean of the x-ray photon energies within the beam.

Radiation shielding garments are commonly used in hospitals, clinics and dental offices to protect medical patients and workers from unintentional direct and secondary radiation exposure during diagnostic imaging. To Decrease the Exposure of Ionizing Radiation in this research discusses this problem. The use of composite materials is an effective methods for altering the performance of Polymer in engineering plastics, rubber, and fiber materials. Composite materials are used for many reasons Such as lowering cost the Industrial, Shielding Protection, In this study, 8 different rubber compound were prepared by using (SBR) with Zinc Oxide (ZnO) at different ratio (10,30,40,50,60,70,80 and 100) pphr and constant ratio of Titanium Dioxide (TiO_2) at 60 pphr. The physical properties such as Tensile, Elongation, Young Modulus, Compression and (X-ray) Transmission were Studied.

Key words: X-ray beam, effective energy, ZnO , SBR, Radiation.

Introduction

In principles, the basic properties of the elastomers come from its nature. However the incorporation of other ingredient into the matrix can modify the properties. The process of the introducing the chemicals or additives into the rubber to modify its properties is called rubber compounding[1]. A good compounding needs to consider many aspects such as environmentally safe, good process ability, satisfactory service life and minimum production cost. The different types of additives and chemical contribute to the above factors [2]. The early stage of the rubber compounding is the softening process of raw rubber by mastication. Sometimes the proptisers will be added. In rubber industry, the widely used equipment to masticate rubber is two roll mills[3]. The mastication is normally applied to SBR. The synthetic rubber seldom needs prior mastication because they are tailored made and can be processed directly[1]. The mastication time of SBR is longer than synthetic rubber because the SBR is normally supplied in high Mooney Viscosity. Mastication time of NR is normally controlled within 15 minutes whereas the synthetic rubber is just only 2 minutes [2]. Mastication process is also important to produce a homogenous dispersion of filler into the rubber matrix. The filler can be only dispersed well in rubber matrix when certain viscosity is achieved. The proper viscosity can improve the processibility of rubber compounds. Basically, there are two categories of mastication process, i.e mastication without peptizes and mastication with peptisers [3]. The mastication without peptizes requires high shear force of two roll mills or internal mixer to break down the polymer chain and consequently reduce the molecular weight[4]. The mastication process depends on the temperature. As the temperature increase, the elastomers soften and consequently absorb less mechanical energy. This is due to the polymer molecules can flow more easily[5]. The high temperature can cause oxidative attack and increase the rate of chain scission consequently reducing the viscosity. The mechanical degradation will then occur and lead to the excessive Vulcanization is a process that increases the overall elasticity of rubber by locking the chains to each other through chemical crosslink's. The slippage behavior of the plastic-like material would change to more dimensional stable material[6]. The most common use of crosslink agent in rubber is sulfur because it is inexpensive and plentiful. This crosslinker can link the double bonds of the rubber chains together. NR and SBR are always crosslinked by this type of vulcanization process due to the only small amount of sulfur to be used[5].

Generally, there are a number of sites which are attractive to sulfur atoms along the rubber molecules called cure sites [6]. In the vulcanization reaction, the eightmembered ring of sulfur breaks down in smaller parts with varying numbers of sulfur atoms. Figure (1) represents the sulfur crosslinking process of polyisoprene. One or

more sulfur atoms can attach itself to the double bond, and then the sulfur can grow until it reaches the other cure sites of double bonds [4]. The sulfur bridge can vary from two to ten atoms[8]. The length of the sulfur chain can affect the physical properties of the vulcanizate. The shorter the sulfur crosslink give the better heat resistance to rubber vulcanizate. Thus, the EV vulcanization system which has lower polysulfide crosslinks gives better heat and aging resistance. However, the high crosslink in the rubber vulcanizate produce very good dynamic properties. The dynamic properties are important in tyre side wall industry [6]. Good flexing properties can reduce the formation of cracks and consequently minimize the failure of the rubber products[5] .

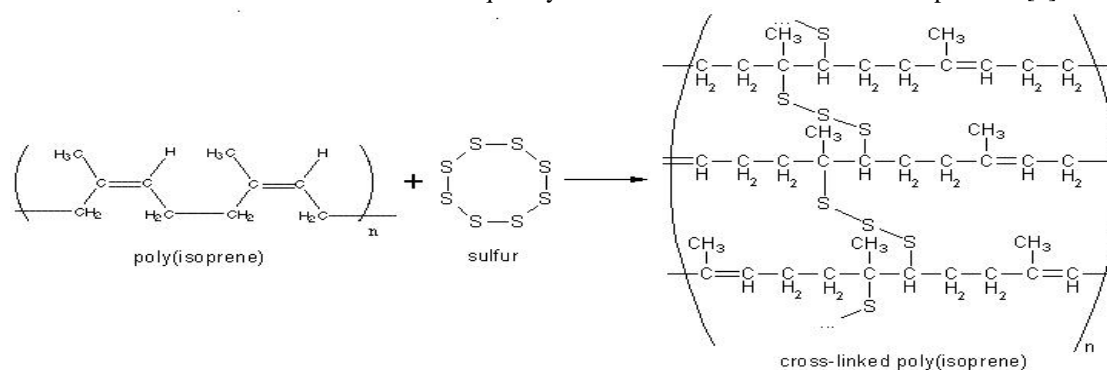


Fig.(1) The sulfur crosslinking process of polyisoprene

Three categories of sulfur vulcanization system are used in rubber technology,i.e conventional vulcanization (CV), semi-efficient vulcanization (semi-EV) and efficient vulcanization (EV). The difference between these systems is the ratio of sulfur and accelerator added into the rubber compounds. Different properties of the vulcanizate can be obtained by varying the sulfur to accelerator ratio in recipe [7].

Experimental

All materials are used in this research come from Babylon Factory Tire Manufacturing , Iraq . The structure of materials is as follows .

* Styrene-butadiene rubber (SBR) . with styrene content 23.5 % , Moony viscosity at 100°C = 50 , specific gravity 0.94 (gm/cm³), ash content 1 % . there are two types of E-SBR in the market . One of them is the hot rubber which is product at 150 °C , Whereby the molecular weight is high and depolymerization Can occur at high temperature [8]. SBR was first discovered by E. Tchunkur and A. Bock by an emulsionpolymerization process in 1929. These emulsion SBR (E-SBR) called Buna S, is easierfor processing compare to other Buna grades SBR another type of E-SBR , cold rubber is using aredox initiator to lower the polymerization tempreature to 5°C and the chain modifier is applied to control the molecular weight [9].

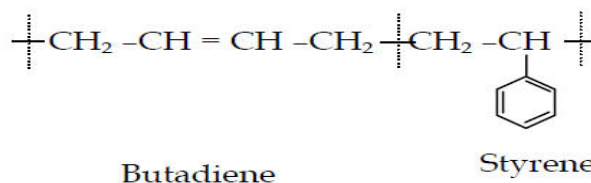


Fig. (2)the chemical formula of SBR [9]

* Titanium Dioxide (TiO₂) is found in abundance in nature as the minerals lmenite (FeTiO₃), rutile (TiO₂) ,and sphere (CaSiTiO₅) among other . the Theoretical density of (TiO₂) ranges from 3895 Kg/m³ for anatase to 4250 Kg/m³ for rutile. The molecular weight is 79.865 , melting point 1843°C , Four naturally occurring titanium dioxide polymorphs exist : rutile ,anatase , brookite and titanium dioxide . Anatase and rutile are tetragonal boorkite is orthormbic and titanium is monoclinic [10].

In all four polymorphs, titanium is coordinated octahedral by oxygen, but the position of octahedral differs between polymorphs. titanium dioxide has also been produced as engineered nanomaterial, which may be equidimensional crystals or sheet and composed of either titanium dioxide – rutile or titanium dioxide – anatase [8].

A tubular structure has been produced from scrolling layers of titanium dioxide – anatase, which result in fibers with an outer diameter of about 6 nm and inner of about 3 nm. Non-scralled nanofibers have also been produced from (TiO₂) "anatase" and (TiO₂) with diameter of 20-100 nm and length of (10-100 μm) [7].

- Antioxidant (6PPD) is a materials of composition [N-(1,3-dimethylbutyl)-N- phynel-P-phenlenediamine] :specific gravity 1.0 (gm/cm³) [9].
- Sulfur : Pale yellow powder of sulfur element, purity 99.0% , melting point 112°C . specific gravity 2.04-2.06 (gm/cm³) [10].
- Zinc Oxide : fine powder , purity 99% , specific gravity 5.6 (gm/cm³) [11].

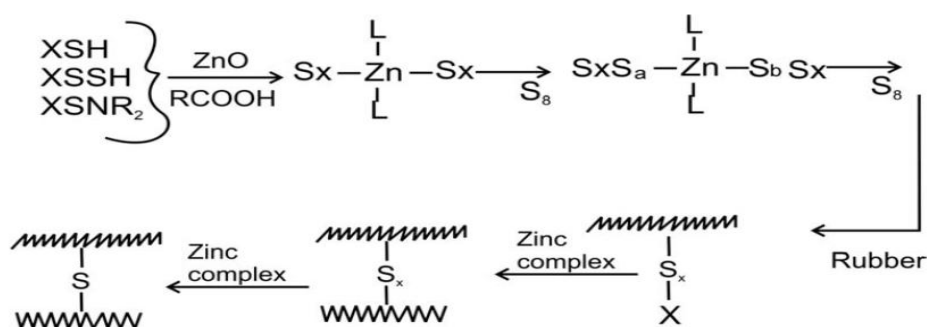


Fig.(3)(Rle of ZnO and fatty acid in accelerated sulfur Vulcanization where X=Accelerator residue , L= ligand) [11].

- Steric acid : melting point 67-69 °C , specific gravity 0.838 (gm/cm³) [8].

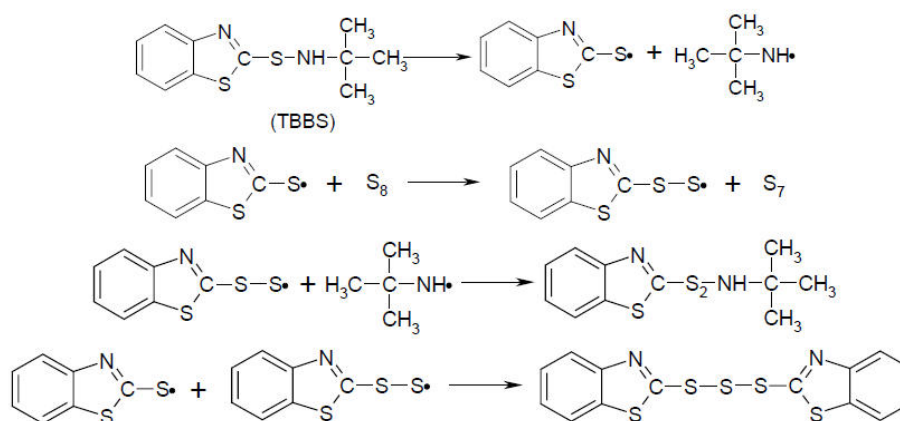


Fig.(4) Reactional pathway the accelerator and sulfur in the absence of an activator [12]

Table (1)the chemical composition for rubber recipe[12]

Compounding ingredients	pphr
Rubber SBR	100
(TiO ₂)	60
ZnO	Variable
TMTD	0.6
Sulfur	2
Satiric Acid	1.5

Results And Discussion

Many tests is carried on to diffine the extent of The addition effect of the different of (ZnO on the properties of (SBR) rubber ,such of this test are:

- **Tensile Test :**

This test is caeeied on according to ASTM D-471-57T specification . The test result for tensile strength are shown in Figure (5). It is seen tag the tensile strength of composite increases with increase in value of (ZnO) because with the grain of titanium dioxide sufface area and it have not the same shape make it to increase cross linking with rubber chain deal to surface area is increase . But we see at the last percent (100 %) that is decrease because the matrix can not having the filler (ZnO) [13].

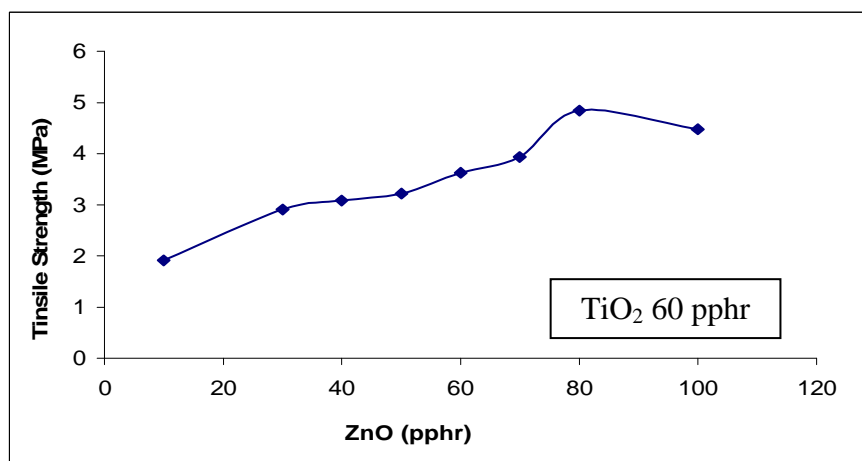


Fig. (5)Effect of (ZnO) on the SBR Tensile

Elongation Test:

This test is caeeied on according to ASTM D-471-57T specification . The test result for Elongation are shown in Figure (6) It is seen that the Elongation increase with percent of (ZnO) at the first three value and become decrease because physical interaction between the (ZnO) and rubber chain , when the grain of filler resistance Elongation [14].

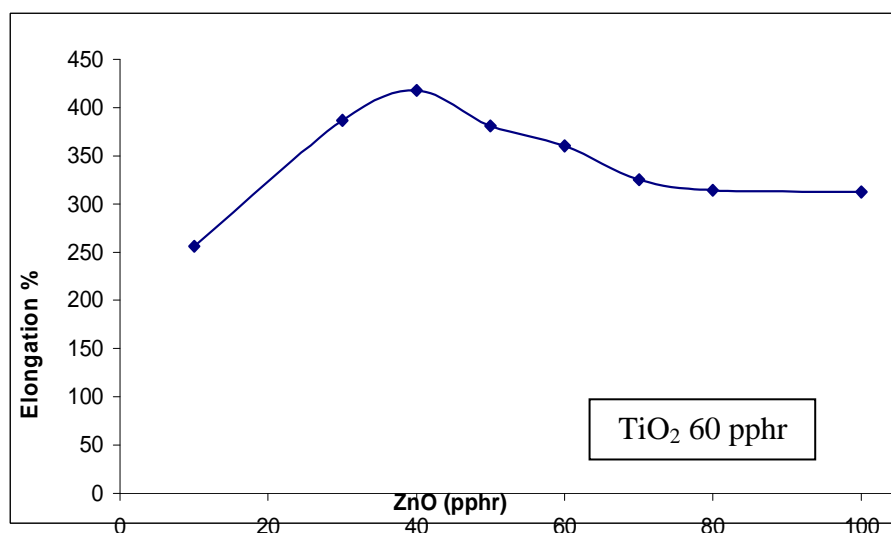
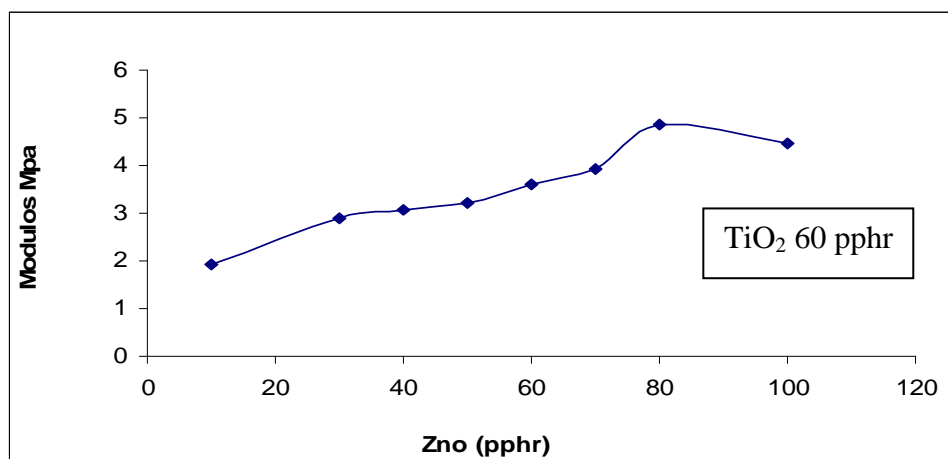


Fig. (6) Effect of (ZnO) on the SBR Elongation

- **Elastic Modulus Test :**

Figure (7) shown a linear increasing between elastic modulus versus (ZnO) weight percent . this is due to thy decomposition of phenol tack resins which acts suitable fillers , phenol Espedialy when combined with suitable filler , phenol resins have good chemical and thermal resistance , dielectric strength and dimension stability. Thus addition of phenol results in decreasing of the elongation and increasing of the tensile strength thus the modulas of elasticity increases[15].



- **Hardness Test :**

Figure (8) shown the shore hardness is plotted against the loading level of reinforcing filler (ZnO) for SBR respectively. From this figure it can be seen that rubber hardness shows signifaction increment with the increasing loading level of renifocing of (ZnO) [15].

Titanium dioxide reinforcing filler have fine graen size , this mean that (ZnO) has larger surface area , which in contact with rubber mostly by physical bond composite with strong bond made it harder by impeding the matrix motion along the stress direction . [16].

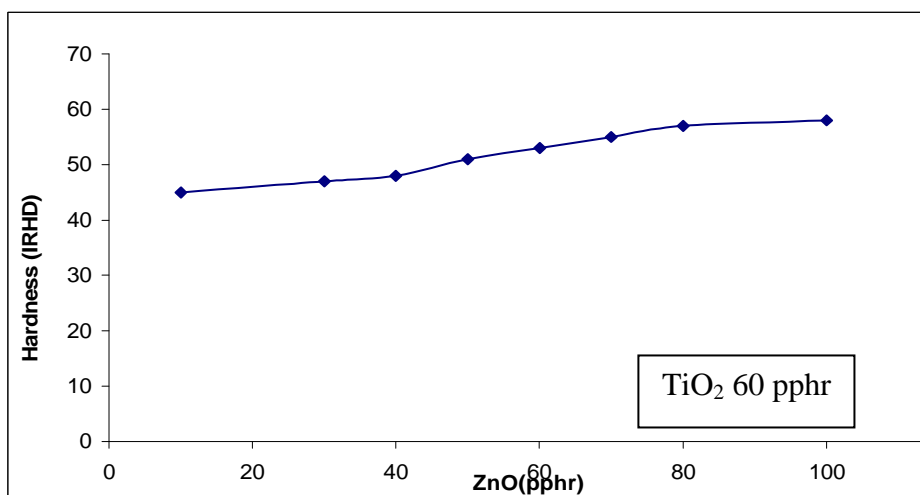


Fig. (8) Effect of (ZnO) on the SBR Hardness

- Resonance Test :**

The relation between Resonance and hardness is invers relation , from figure (9) show the resonance decrease when (ZnO) percent increase , because the cross linking between rubber chain that absorb energy and transford it to heat among the rubber chain . value of resonance decrease when hardness or cross linking increase [17] .

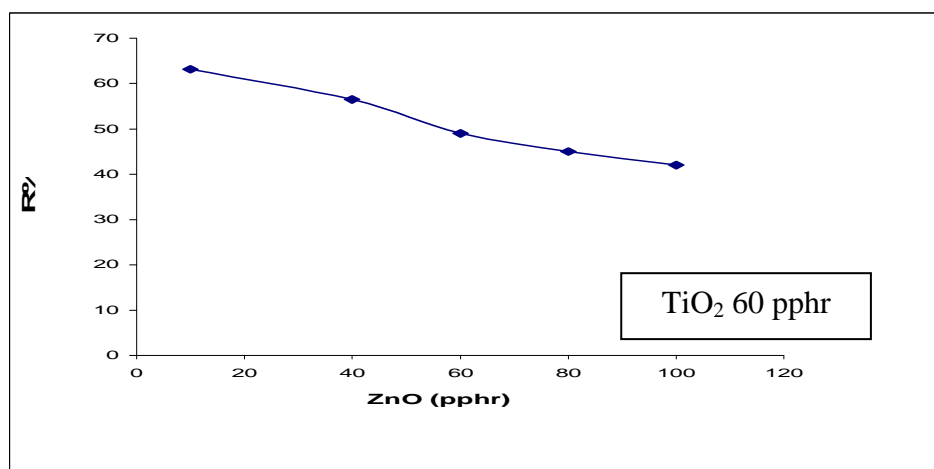


Fig. (9) Effect of (ZnO) on the SBR Resonance

- Compression Test:**

This test is caeeied on according to ASTM D-471-57T specification . The test result for Compression are shown in Figure (10). Because interaction between filler (titanium Dioxide) and rubber (SBR) that lead to increasing of cross linking at 3-dim . (ZnO) properties same grain size that mean it have large surface area helped it to connected with all chain polymer and resistance the load and pressure instead of covalent bond or hydrogen bond when keep surface with out Buckling [18].

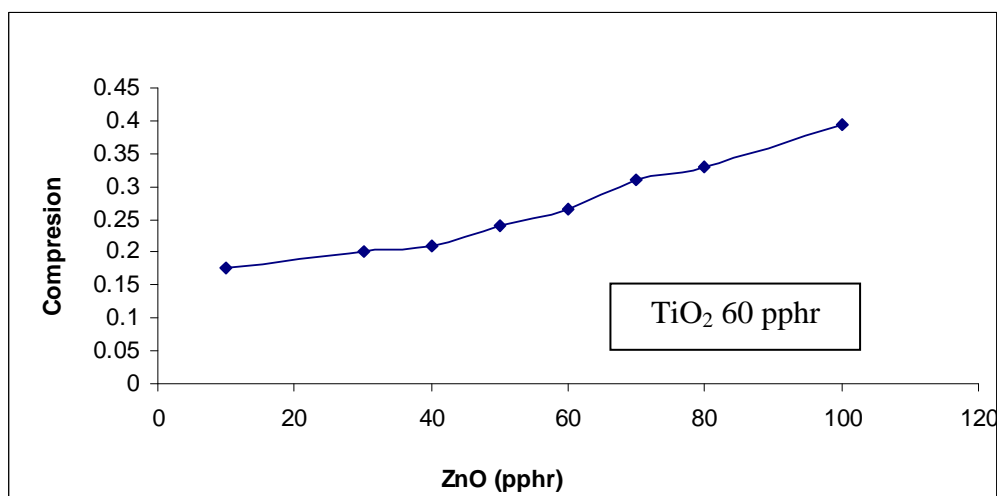
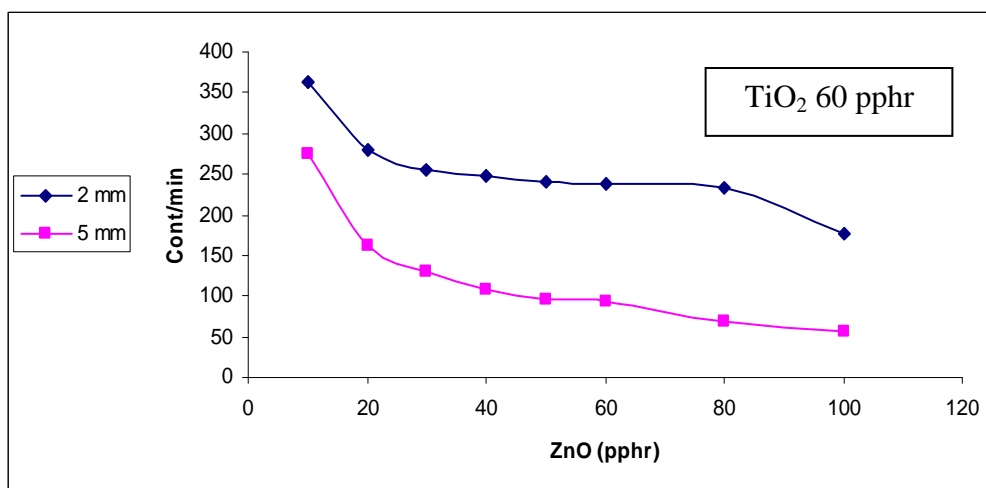


Fig. (10) Effect of (TiO₂) on the SBR Compression

• (X-Ray) Transmmision Test:

Titanium Dioxide is especially efficient to absorb X-Ray in the 30-120 KeV range because of the electronic structure . the molecular weight is 79.865 depending on the application , many artificial structure incorporating lead dioxide , have thus be can used to absorb X-Ray , for example this investigation rubber gloves filled with lead dioxide powder are used to insure agood protection to operators exposed to ionizing radiation hospital. Figure (11) shown that (X-Ray) contract when increasing of (ZnO) value where (ZnO) enjoy from some properties such as absorb and scattering (X-Ray) and interaction between materials led to increasing of composite materials to contract X-ray[16,17]



Conclusion

- 1- Transmission of (X-Ray) that increasing with loading of (ZnO) percent .
- 2- Cross – Linking between compound material that increasing (X-Ray) transmission .

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